



ANURAG GROUP OF INSTITUTIONS

(AUTONOMOUS)

DEPARTMENT OF MECHANICAL ENGINEERING

II B.TECH – I SEM- I ASSIGNMENT EXAMINATION

THERMODYNAMICS

Time: 60 Min

Max.marks: 5

Academic year: 2018-19

1. Explain in detail about concept of continuum with neat sketch?
 - a) A quantity of air has a volume of 2.5m^3 and a pressure of 12bar at temperature 135°C . It is expanded in a cylinder at a constant temperature to a pressure of 3.8bar. Determine the amount of work done by the air during expansion?
2. Differentiate between path and point functions? Are work and heat path functions? Justify.
 - a) Derive the work done equations for polytropic non-flow process in a closed system?
3. define universe, control volume, thermodynamic system and boundary, steady flow and unsteady flow process?
4.
 - a) Differentiate macroscopic view and microscopic view.
 - b) Explain the constant volume gas thermometer with neat sketch.
5.
 - a) Derive steady flow energy equation for throttling device and steam condenser?
 - b) A gas expands through an ideally insulated nozzle following a reversible polytropic law $pv^{1.2}=c$. There is no change in potential energy but the pressure drops from 15bar to 6.2bar and the specific volume increases from 1.5m^3 to 4.3m^3 . if the velocity is 80 m/sec. determine the exit velocity.
6.
 - a) Explain joules experiment for a closed system undergoing a cyclic process with neat sketch?
 - b) A stationary mass of gas is compressed without friction from an initial state of 5m^3 and 10Mpa to a final state of 2m^3 , remaining pressure as constant during the process. There is a transfer of 170KJ of heat from the gas during the process. Calculate the internal energy during the change of process?